

Tandem Diels–Alder/Cross-Coupling Reactions of Generated in Situ Organoindium Reagents in a One-Pot Process

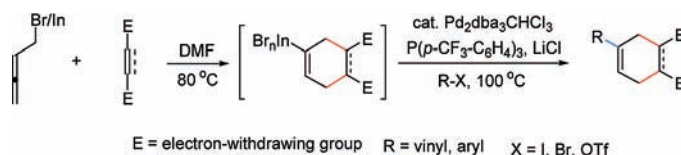
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Received November 2, 2009

ABSTRACT



Tandem Diels–Alder/ cross-coupling reactions with organoindium reagents generated in situ from 1-bromo-2,3-butadiene and indium were developed in a one-pot process. [4 + 2] Cycloaddition reactions using organoindium reagents and subsequent Pd-catalyzed cross-coupling reactions provided the rapid synthesis of six-membered carbocycles starting from 1-bromo-2,3-butadiene.

Because main group element substituted 1,3-dienes and their synthons play an important role as building blocks in organic synthesis, development of efficient synthetic methods for their preparation has been required. Although a number of 1-main group element substituted 1,3-dienes were reported,¹ 2-main group element substituted 1,3-dienes are rather difficult to prepare and, therefore, rarely reported. In particular, of these compounds, 2-(1,3-butadienyl)magnesium chloride has been known to be a tricky species mainly due to the poor regioselectivity involved during the course of the reactions.² While synthesis of 2-phenylseleno- and 2-trialkylstannyl-1,3-butadiene was reported by Bates et al.,³ 1,3-butadienyl-2-ylolithium was indirectly prepared from the

2-stannyl precursor.⁴ Preparation of 2-trialkylsilyl-1,3-butadiene and 2-silyl-substituted 1,3-cyclohexadienes was also reported.⁵ Eisch and Hoberg reported previously about aluminum-substituted 1,3-dienes.⁶ In addition, 1,3-dienyl-2-zirconium compounds were prepared from the reaction of 2-trimethylsilyloxy-1,3-dienes with zirconocene.⁷ Recently, Welker et al. reported the preparation of 2-trialkylsilyloxy-substituted and 2-KBF₃-substituted 1,3-dienes from the treatment of 2-chloro-1,3-butadiene with Mg followed by trimethyl boronate and tri(ethoxysilyl) chloride, respectively.⁸

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However, synthons for 2-main group element substituted 1,3-dienes such as homoallyl silane,⁹ stannane,¹⁰ and boronate¹¹ are inconvenient to prepare. Because so much is now known about the Diels–Alder reaction,¹² we were convinced that when we developed an efficient synthon for 2-main group element substituted 1,3-dienes, this would prove useful for the synthesis of six-membered carbocycles bearing a vinyl metal moiety. Recently, we discovered that organoindium reagents generated in situ from the reaction of indium with 1-bromo-2,3-butadiene or 1,3-dibromo-2-butyne acted as efficient synthons for 1,3-dienyl-2-yl species in Pd-catalyzed cross-coupling reactions¹³ and addition reactions to imines and aldehydes.¹⁴ During the course of our research program aimed at finding new indium-mediated organic reactions,¹⁵ we envisioned the possibility of extending the tandem reaction by using an easily accessible homoallyl indium. Herein, we report tandem Diels–Alder/cross-coupling reactions with organoindium reagents generated in situ from 1-bromo-2,3-butadiene and indium in a one-pot process, producing cyclohexene derivatives via formation of 1-cyclohexenyl-1-indium (Scheme 1).

Scheme 1. Tandem Diels–Alder Reaction/Cross-Coupling Reaction

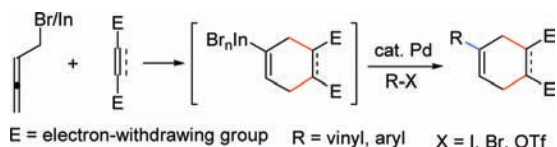
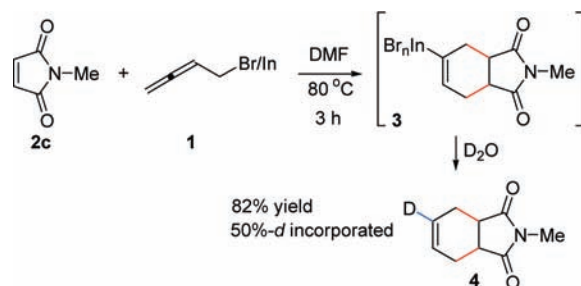


Table 1. Diels–Alder Reaction of Organoindium with Dienophile^a

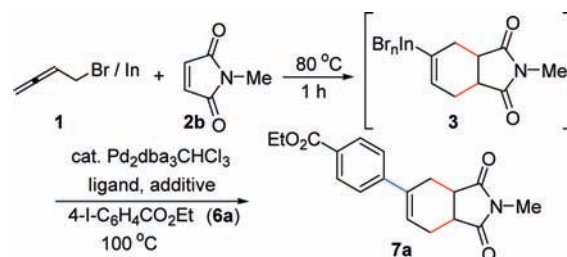
entry	dienophile	adduct	time (h)	yield (%)
1 ^b			3	83
2			2	82 ^c
3 ^b			1	76 ^c (16) ^d
4			2	83 ^c
5			2	84

^a **1** (0.75 mmol), In (0.5 mmol), and **2** (0.5 mmol) in DMF (0.25 M) was used. ^b 2 equiv of **2** was used. ^c Cis. ^d 2.



First, we examined whether organoindium reagents generated in situ from 1-bromo-2,3-butadiene and indium could work as an efficient partner suitable for a Diels–Alder reaction with dienophiles (Table 1). When *N*-methylmaleimide **2b** (0.5 mmol) was treated with organoindium reagent, generated in situ from 1-bromo-2,3-butadiene **1** (0.75 mmol) and indium (0.5 mmol) in DMF, at 80 °C for 2 h, tetrahydroisindole-1,3-dione **5b** was obtained in 82% yield (entry 2). This result implies that the organoindium reagent acted efficiently as 1,3-butadienyl-2-ylindium. Quenching of the reaction mixture with D₂O after the initial Diels–Alder reaction gave rise to the corresponding deuterated adduct **4** in 82% yield with 50% *d*-incorporation, indicating that 1-cyclohexenyl-1-indium **3** was indeed

Table 2. Optimization of Tandem Diels–Alder/Cross-Coupling Reaction^a



entry	ligand	additive	solvent	time (h)	yield ^b (%)	
					5b	7a
1	16 PPh ₃	LiI	DMF	2	26 ^c	48 ^c
2	8 Xantphos	LiI	DMF	3	13	43
3	8 DPEphos	LiI	DMF	3	6	33
4	8 DPPF	LiI	DMF	2	5	47
5	16 P(2-furyl) ₃	LiI	DMF	2	11	69 (64) ^c
6	16 (biphenyl)Pcy ₂	LiI	DMF	2	13	31
7	16 P(C ₆ F ₅) ₃	LiI	DMF	2	5	66 (60) ^c
8	16 P(<i>p</i> -MeO-C ₆ H ₄) ₃	LiI	DMF	3		26
9	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃		DMF	2		52 ^c
10	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiI	DMF	2		70 ^c
11	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiI ^d	DMF	2		5
12	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiI	THF	2	65	28
13	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiI	CH ₃ CN	3	7	56 (54) ^c
14 ^e	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiI	DMF	2	8	92 (89) ^c
15 ^e	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiCl	DMF	2		93 ^c
16 ^e	16 P(<i>p</i> -CF ₃ -C ₆ H ₄) ₃	LiBr	DMF	2		84 ^c

^a 2 mol % of Pd₂dba₃CHCl₃, **1** (1.5 equiv), **6a** (1 equiv), In (1 equiv), **2b** (1.5 equiv), and additive (1 equiv) were used. Stereochemistry of **7a** is cis. ^b GC yield based on C₁₄H₃₀ as an internal standard. ^c Isolated yield. ^d LiI (2 equiv) was used. ^e **1** (2.3 equiv) was used.

formed. Next, we applied these conditions to a variety of dienophiles. *N*-Phenylmaleimide afforded the corresponding adduct (**5c**) in 76% yield under the optimum reaction conditions (entry 3). DMAD reacted with the organotin reagent in DMF at 80 °C for 3 h, producing 1,2-

bis(methoxycarbonyl)-1,4-cyclohexadiene **5a** in 83% yield (entry 1). We were particularly pleased to obtain cyclohexene adducts in 83% and 84% yields, respectively, from the reaction with maleic anhydride and ethyl acrylate (entries 4 and 5).

Table 3. Tandem Diels–Alder/Cross-Coupling Reaction in a One-Pot Process^a

entry	dienophile	electrophile	product	time (h)	yield (%)
1				6	0
2				3	76
3		X = Cl		2	93
4		Br		4	72
		I			
		OTf			
5	2b			2	81
6	2b			2	88 (1:1.3) ^c
7	2b			3	72
8	2b			2	60
9 ^d				3	79
10	2c			2	73
11 ^e				3	60
12 ^e	2f			2	78
13 ^d				2	68
14				2	53
15				2	86(1.4) ^g

^a Reaction were carried out with **1** (1.1 mmol), In (0.75 mmol), and **6** (0.5 mmol) in DMF (0.25 M). ^b Cis. ^c Diastereomeric ratio. ^d **2** (3 equiv) was used. ^e Reactions were carried out with **1** (1.5 mmol), In (1.0 mmol), In (1.0 mmol), **2** (1.0 mmol), and **6** (0.5 mmol). ^f Trans. ^g Ratio of *meta*- and *para*-isomers.

Next, detection of cyclohexenylindium as an intermediate led us to investigate tandem Diels–Alder/ cross-coupling reactions using 1-bromo-2,3-butadiene and indium in a one-pot process. These results are summarized in Table 2. After *N*-methylmaleimide **2b** reacted with organoindium reagents in DMF, the reaction mixture was treated with ethyl 4-iodobenzoate in the presence of a variety of Pd catalysts to provoke tandem Diels–Alder/cross-coupling reactions. Of the catalytic systems examined, the best results were obtained with 2 mol % of Pd₂dba₃CHCl₃ and 16 mol % of (*p*-CF₃-C₆H₄)₃P in the presence of 1 equiv of LiCl in DMF at 100 °C for 2 h under a nitrogen atmosphere, affording **7a** selectively in 93% yield (entry 15). When Ph₃P, Xantphos, or (biphenyl)PCy₂ was used, the cross-coupling reactions did not completely proceed (entries 1, 2, and 6). The use of (*p*-MeO-C₆H₄)₃P provided tandem reaction product in only 26% yield (entry 8). THF gave the desired product **7a** in 28% yield together with Diels–Alder adduct **5b** in 65% yield (entry 12). Of the additives examined, LiCl (1 equiv) gave the best results.

To demonstrate the efficiency and scope of the present method, we applied this catalytic system to a variety of dienophiles and electrophilic coupling partners in the tandem Diels–Alder/cross-coupling reactions. These results are summarized in Table 3. Although ethyl 4-chlorobenzoate did not react with cyclohexenylindium derived from *N*-methylmaleimide, 1-bromo-2,3-butadiene, and indium under the optimum reaction conditions (entry 1), the corresponding bromide and triflate produced the desired tandem reaction products **7a** in 76% and 72% yields, respectively (entries 2

and 4). Exposure of cyclohexenylindium to vinyl bromide and vinyl triflate afforded **7b** and **7c** in 81% and 88% (dr = 1:1.3) yields, respectively (entries 5 and 6). The present method worked equally well with diverse aryl iodides such as 4-iodo-*n*-butylbenzene and 2-iodothiophene (entries 7 and 8). Cyclohexenylindium generated in situ from *N*-phenylmaleimide reacted efficiently with 2-iodobenzaldehyde and 3-iodoaniline, producing tandem Diels–Alder/cross-coupling products (**7f** and **7g**) possessing a 2-formylphenyl and a 3-aminophenyl group in 79% and 73% yields (entries 9 and 10). Vinylindium generated in situ from dimethyl fumarate coupled with 2-iodoanisole, affording tandem reaction products (**7h**) in 60% yield (entry 11). 3-Iodopyridine turned out to be compatible with the present reaction conditions, producing adduct **7i** in 78% yield (entry 12). Treatment of cyclohexenylindium generated in situ from DMAD and maleic anhydride with 4-iodo-*n*-butylbenzene and ethyl 4-iodobenzoate provided **7j** and **7k** in 68% and 53% yields, respectively, in a one-pot process (entries 13 and 14). We were pleased to obtain the functionalized cyclohexene **7l** in 86% yield from the reaction of ethyl acrylate with ethyl 4-iodobenzoate (entry 15).

In summary, we have developed an efficient new tandem procedure of Diels–Alder/cross-coupling reactions with organoindium reagents generated in situ from 1-bromo-2,3-butadiene and indium in a one-pot process. [4 + 2] Cycloaddition reactions of organoindium reagents with various dienophiles and subsequent Pd-catalyzed cross-coupling reactions provided the rapid synthesis of six-membered carbocycles starting from aryl bromides and iodides, vinyl bromides and triflates, 1-bromo-2,3-butadiene, and dienophiles.

Acknowledgment. This work is dedicated to Professor Bong Young Chung on the occasion of his honorable retirement from Korea University. This work was supported by the NRF through the NRL Program (No. M10600000203-06J0000-20310), by the KRF Grant funded by the Korean government (KRF-2008-314-C00213) by the NRF grant funded by the Korea government (MEST) (2009-0087013), and by Korea Sanhak Foundation. This work was supported by the second phase of the BK 21 Program in 2009. The NMR data were obtained from the central instrumental facility in Kangwon National University. We thank Professor S. Chang of KAIST and Professor M. E. Welker of Wake Forest University for proofreading this manuscript.

Supporting Information Available: Experimental procedure and spectral data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL902530K

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